

Achievement Level Descriptors for Physical Science

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Achievement Levels and Achievement Level Descriptors

With the implementation of the Georgia Milestones Assessment System, Georgia educators have developed four achievement levels to describe student mastery and command of the knowledge and skills outlined in Georgia's content standards. Most students have at least some knowledge of the content described in the content standards; however, achievement levels succinctly describe how much mastery a student has. Achievement levels give meaning and context to scale scores by describing the knowledge and skills students must demonstrate to achieve each level.

The four achievement levels on Georgia Milestones are *Beginning Learner*, *Developing Learner*, *Proficient Learner*, and *Distinguished Learner*. The general meaning of each of the four levels is provided below:

Beginning Learners do not yet demonstrate proficiency in the knowledge and skills necessary at this grade level/course of learning, as specified in Georgia's content standards. The students *need substantial academic support* to be prepared for the next grade level or course and to be on track for college and career readiness.

Developing Learners demonstrate partial proficiency in the knowledge and skills necessary at this grade level/course of learning, as specified in Georgia's content standards. The students *need additional academic support* to ensure success in the next grade level or course and to be on track for college and career readiness.

Proficient Learners demonstrate proficiency in the knowledge and skills necessary at this grade level/course of learning, as specified in Georgia's content standards. The students are prepared for the next grade level or course and are on track for college and career readiness.

Distinguished Learners demonstrate advanced proficiency in the knowledge and skills necessary at this grade level/course of learning, as specified in Georgia's content standards. The students *are well prepared* for the next grade level or course and are well prepared for college and career readiness.

More detailed and content-specific concepts and skills are provided for each grade, content area, and course in the **Achievement Level Descriptors** (ALDs). ALDs are narrative descriptions of the knowledge and skills expected at each of the four achievement levels and were developed for each grade level, content area, and course by committees of Georgia educators in March 2015 and July 2015. The ALDs are based on the state-adopted content standards.

ALDs show a *progression of knowledge and skills* for which students must demonstrate competency across the achievement levels. It is important to understand that a student should demonstrate mastery of the knowledge and skills within his/her achievement level *as well as all content and skills in any achievement levels that precede his/her own, if any*. For example, a Proficient Learner should also possess the knowledge and skills of a Developing Learner *and* a Beginning Learner.

POLICY ALDs			
Beginning Learner	Developing Learner	Proficient Learner	Distinguished Learner
Beginning Learners do not yet	Developing Learners demonstrate	Proficient Learners demonstrate	Distinguished Learners
demonstrate proficiency in the	partial proficiency in the	proficiency in the knowledge and	demonstrate advanced
knowledge and skills necessary at	knowledge and skills necessary at	skills necessary at this grade	proficiency in the knowledge and
this grade level/course of learning,	this grade level/course of learning,	level/course of learning, as	skills necessary at this grade
as specified in Georgia's content	as specified in Georgia's content	specified in Georgia's content	level/course of learning, as
standards. The students need	standards. The students need	standards. The students are	specified in Georgia's content
substantial academic support to be	additional academic support to	prepared for the next grade level or	standards. The students are well
prepared for the next grade level or	ensure success in the next grade	course and are on track for <i>college</i>	prepared for the next grade level
course and to be on track for	level or course and to be on track	and career readiness.	or course and are well prepared
college and career readiness.	for college and career readiness.		for college and career readiness.
RANGE ALDs			
Beginning Learner	Developing Learner	Proficient Learner	Distinguished Learner
A student who achieves at the	A student who achieves at the	A student who achieves at the	A student who achieves at the
Beginning Learner level	Developing Learner level	Proficient Learner level	Distinguished Learner level
demonstrates minimal command of	demonstrates partial command of	demonstrates proficiency of the	demonstrates advanced
the course standards. The pattern	the course standards. The pattern	course standards. The pattern	proficiency of the course
exhibited by student responses	exhibited by student responses	exhibited by student responses	standards. The pattern exhibited
indicates that students are most	indicates that students are most	indicates that students are most	by student responses indicates
likely able to	likely able to	likely able to	that students are most likely able
 recognize the differences 	identify the structure of the	examine the structure of the	to
between atoms and molecules;	atom;	atom in terms of proton,	 describe nuclear energy, its
 describe the movement of 	 recognize that electrons are 	electron, and neutron locations,	practical application as an
particles in solids, liquids, gases,	involved in bonding;	atomic mass and atomic	alternative energy source, and
and plasmas;	 compare and contrast the 	number, and atoms with	its potential problems;
 use the Periodic Table of the 	atomic/molecular motion of	different numbers of neutrons	differentiate between alpha
Elements to locate metals,	solids, liquids, gases, and	(isotopes) and explain the	and beta particles and gamma
nonmetals, and metalloids;	plasmas;	relationship of the proton	radiation;
 describe the Law of 	explore the nature of matter, its	number to the element's	 predict formulas for stable
Conservation of Matter;	classifications, and the system	identity;	binary ionic compounds based
 describe energy 	for naming types of matter;	compare and contrast ionic and	on balance of charges;
transformations;	describe the three types of heat	covalent bonds in terms of	 describe solutions in terms of
 explain that colors are 	energy transfer (radiation,	electron movement;	concentration and
distinguished by differences in	conduction, convection);	identify the types of	conductivity;
wavelengths/frequencies;	investigate the properties of	radioactivity and differentiate	 demonstrate that solubility is

- describe how the behavior of waves is affected by medium (air, liquid water, solids);
- explain how the parts of a wave are affected by changes in amplitude and pitch;
- demonstrate series and parallel circuits;
- identify three types of heat transfer;
- recognize simple machines that make work easier;
- recognize appropriate laboratory techniques;
- suggest reasonable hypotheses for identified problems;
- recognize possible effects of measurement and calculation errors;
- identify reactants and products in a chemical reaction;
- identify simple machines;
- recognize examples of forces; and
- recognize that many electrons produce electricity.

- solutions by observing factors affecting the rate at which a solute dissolves in a specific solvent;
- identify phase changes;
- identify energy transformations within a system, such as lighting a match;
- recognize that all waves transfer energy;
- recognize appropriate solutions for solving scientific problems;
- use technology to develop tables and graphs;
- use dimensional analysis to solve problems;
- use data as evidence to support scientific claims;
- recognize a balanced equation;
- recognize balanced and unbalanced forces;
- describe electromagnetic and mechanical waves;
- identify methods of electron transfer;
- define voltage, resistance, and current;
- identify types of circuits; and
- describe a permanent magnet.

- between fission and fusion;
- explain the process of half-life as it relates to radioactive decay;
- demonstrate the effect of balanced and unbalanced forces on an object;
- determine the trends in the number of valence electrons, types of ions formed by representative elements, and phases at room temperature;
- use the Periodic Table of the Elements to predict the properties for representative elements;
- relate temperature, pressure, and volume of gases to the behavior of gases;
- use IUPAC nomenclature for transitions between chemical names and chemical formulas of binary ionic compounds (containing representative elements) and binary covalent compounds (carbon dioxide, carbon tetrachloride);
- apply the Law of Conservation of Matter in a chemical reaction;
- calculate density when given a means to determine a substance's mass and volume;
- balance chemical equations (synthesis, decomposition, single replacement, double

- related to temperature by constructing a solubility curve;
- describe molecular motion as it relates to thermal energy changes in terms of conduction, convection, and radiation;
- explain the flow of energy in phase changes through the use of a phase diagram;
- explain magnetism and/or its relationship to the movement of electrical charge as it relates to electromagnets, including simple motors and permanent magnets;
- solve scientific problems by substituting quantitative values or using dimensional analysis and/or simple algebraic formulas as appropriate;
- determine the half-life given a graph of radioactive decay;
- relate frequency and wavelength to the energy of different types of electromagnetic waves and mechanical waves;
- compare and contrast the characteristics of electromagnetic and mechanical (sound) waves;
- predict which machine would have the greatest mechanical advantage;
- calculate the velocity of a

- replacement);
- describe solutions in terms of solute/solvent;
- compare and contrast the components and properties of acids and bases;
- determine whether common household substances are acidic, basic, or neutral;
- differentiate between conduction, convection, and radiation;
- determine the heat capacity of a substance using mass, specific heat, and temperature;
- compare phase changes;
- calculate velocity and acceleration;
- apply Newton's three laws to everyday situations;
- relate falling objects to gravitational force;
- explain the difference between mass and weight;
- calculate amounts of work and mechanical advantage for simple machines;
- investigate the phenomena of reflection, refraction, interference, and diffraction;
- relate the speed of sound to different mediums;
- explain the Doppler Effect in terms of everyday interactions;
- investigate static electricity in terms of friction, induction, and

- falling object;
- apply alternating and direct current in a real-world scenario;
- solve problems related to voltage, resistance, and current;
- explain how factors affect the strength of an electromagnet;
 and
- predict outcomes given series and parallel circuits.

Physical Science EOC	al Science EOC Georgia End-of-Course: Science	
	conduction; explain the flow of electrons in terms of alternating and direct current; explain the relationships between voltage, resistance, and current; compare and contrast simple series and parallel circuits; describe an electromagnet; recognize that different explanations can be used to explain a single set of data; and use technology to develop and test experimental or mathematical models.	